Seasonal climate signals (1990–2008) in a modern Soreq Cave stalagmite as revealed by high-resolution geochemical analysis

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A R T I C L E   I N F O

Article history:
Received 11 January 2013
Received in revised form 13 November 2013
Accepted 15 November 2013
Available online 25 November 2013

Editor: Michael E. Böttcher

Keywords:
Speleothem
Climate record
Oxygen isotopes
Trace elements
Ion microprobe
Laser ablation ICP-MS

A B S T R A C T

High-resolution isotopic and geochemical analyses in a modern (1990–2008) Soreq Cave stalagmite are compared to instrumental records of rainfall and dripwater from the cave, with the aim of determining how seasonal-resolution climate information is transmitted to speleothem geochemistry. In situ, micron-scale analysis of oxygen isotope ratios (δ18O) and trace elements by ion microprobe in combination with a continuous, linear traverse of trace element concentrations by laser-ablation ICP-MS (LA-ICP-MS) allows the definition of geochemical pathways within the cave. Fluorescent banding, imaged by confocal laser fluorescent microscopy (CLFM), as well as δ18O and trace element variations is used to define 18 annual growth increments. Reduced intensity of fluorescent banding and a change in trace element variability reflect the decrease in average rainfall from 628 mm/yr (1990–1998) to 433 mm/yr (1999–2008). During the wetter period before 1998, Pearson (r-value) and Spearman (p-value) correlation coefficients are >0.5 for ion microprobe analyses of the element pairings Sr–Y, Y–P, and Mn–Sr. After the transition to the drier period in 1999, a different set of geochemical pairings have r- and p-values >0.5, including Mg–δ18O, Mg–Sr, and Sr–Ba. Principal component analysis of data from the adjacent LA-ICP-MS traverse identifies two primary underlying modes of trace element variability. Based on the ion microprobe correlations and principal component analyses, we suggest that a greater seasonal influx of particulate material into the cave during the wetter period (1990–1998) brought about greater P, Cu, Sr, Na, and U variability in the stalagmite. The co-variability of δ18O, Mg, Sr, and Ba is characteristic of the trace element pattern from the drier period (1999–2008) of growth when particulate transport is reduced. These findings support a two-reservoir model of Soreq Cave dripwaters. One reservoir displays a well-mixed "baseline" with a decadal residence time that supplies water to the cave year-round, probably from fine pores or grain-boundary films in the vadose zone. The second reservoir is seasonal rainfall enriched in organic acids, colloids, and small particles and is rapidly transmitted to the cave. Finally, the similar patterns of fluorescence intensity, P, and Cu concentrations support the hypothesis that fluorescent bands in Soreq Cave speleothems are caused by the influx of organic colloids.

1. Introduction

Cave deposits (speleothems) often preserve climate information in their geochemistry. Measurements of oxygen and carbon isotope ratios (δ18O, δ13C) as well as trace element concentrations in carbonate speleothems are used to decipher both paleoclimate and paleoenvironmental conditions (Fairchild et al., 2000; Richards and Dorale, 2003; McDermott, 2004; Fairchild et al., 2006; Baldini et al., 2008; Cheng et al., 2009; Lachniet, 2009; Cheng et al., 2012). The need to improve the continental paleoclimate record, combined with innovations in high spatial-resolution geochemical analyses, has resulted in development of seasonal-resolution climate records from speleothems. Multiple in situ analytical techniques permit seasonal-resolution analyses: laser-ablation inductively-coupled-plasma mass-spectrometry (LA-ICP-MS) and synchrotron radiation micro X-ray fluorescence for trace element analyses; ion microprobe for δ18O and trace element analyses (Fairchild et al., 2001; Finch et al., 2001; Huang et al., 2001; Kuczmow et al., 2001; Treble et al., 2003, 2005; Borsato et al., 2007; Orland et al., 2009; Frisia et al., 2012). High spatial-resolution records of trace elements in speleothems reflect seasonal changes in both cave atmosphere and hydrology that respond to regional climate...
Drill-sampling techniques for δ18O accuracy and precision of sampling at the required spatial resolution. Recent advances in the accuracy and precision of δ18O analyses by the ion microprobe at the WiscSIMS Laboratory, University of Wisconsin-Madison (UW-Madison), allow users to resolve subtle δ18O variability in carbonate paleoclimate proxy records on a 10 μm scale (Weidel et al., 2007; Orland et al., 2009; Valley and Kita, 2009; Kozdon et al., 2011; Orland et al., 2012). With this technique, it is now possible to measure the variability of δ18O in speleothems at a finer spatial resolution than trace element measurements by LA-ICP-MS; seasonal-scale climate interpretations that integrate analyses of δ18O and trace elements are now viable in many more speleothems.

This study combines high-resolution data from several techniques in order to examine the coupled behavior of δ18O and trace elements in a modern speleothem (sample 5b) from the semi-arid Soreq Cave locality in Israel (31°45.35′N, 35°1.35′E; Fig. 1). Measurements by LA-ICP-MS (continuous trace element traverse) and WiscSIMS ion microprobe (trace element and δ18O spot analyses) are paired with imaging by confocal laser fluorescent microscopy (CLFM). The results will provide a basis for paleoclimate interpretations of both δ18O and trace element records from Soreq Cave, which is a well-studied site in the Judean Hills ~400 m above sea level and 40 km inland from the Mediterranean Sea, as well as other caves in semi-arid regions (Bar-Matthews et al., 1997; Ayalon et al., 1998; Kaufman et al., 1998; Ayalon et al., 1999; Bar-Matthews et al., 2003; Kolodny et al., 2003; McGarry et al., 2004; Affek et al., 2008; Bar-Matthews and Ayalon, 2011).

Prior work on Soreq Cave speleothems at WiscSIMS used CLFM imaging of fluorescent bands to guide sub-annual-resolution ion microprobe analyses of δ18O (Orland et al., 2009, 2012). Coupled variability of fluorescence and δ18O values was first identified across what were interpreted as annual growth bands in a Late Holocene speleothem (Orland et al., 2009). In that study, it was proposed that fluorescent bands were caused by organic compounds like humic and fulvic acids that were flushed into the cave during an annual winter wet-season. Orland et al. (2009) show that the sharp onset of each fluorescent band coincides with a drop in speleothem δ18O that resembles the modern seasonal variability in dripwater and rainfall δ18O values; modern values of rainfall δ18O and dripwater δ18O are lowest from November to April when ~95% of annual rainfall occurs (Fig. 2). Cave monitoring at Soreq shows that as the winter wet-season ends, the source of dripwater gradually shifts to a high-δ18O “baseline” vadose-zone reservoir (Ayalon et al., 1998) that is suggested to have fewer organic colloids in suspension (Orland et al., 2009). Based on this evidence, Orland et al. (2009) suggested that regular, sawtooth variability of δ18O and fluorescence observed in Holocene Soreq speleothems mark annual growth bands; the gradual increase of δ18O across each band accompanies a decrease in fluorescence and ends with an abrupt return to low δ18O values at the onset of the next winter wet-season and bright fluorescent band.

Here we seek to further test the hypothesis that coupled variability of fluorescence and δ18O delineate annual growth bands and rainfall amounts in a detailed analysis of modern stalagmite 5–3b. The age of sample 5–3b is constrained by its growth on a man-made object placed in Soreq Cave in 1990; the sample was collected in 2008. Because the maximum age of sample 5–3b is known, we can: (1) test the hypothesis that the fluorescent bands are annual markers, and subsequently (2) compare speleothem geochemistry to instrumental records of both rainfall above Soreq and dripwater within the cave. In addition, we use correlation statistics and principal component analysis to assess the co-variability of δ18O, trace elements, and fluorescence measured in sample 5–3b (e.g. Borsato et al., 2007; Wassenburg et al., 2012). These comparisons allow us to evaluate the response of the geochemistry of 5–3b to changes in annual rainfall amount and test whether organic colloids are the cause of fluorescent banding in the sample.

2. Materials and methods

Sample 5–3b is a stalagmite that formed on the base of a stationary drip-collecting apparatus placed in Soreq Cave in July 1990. The stalagmite grew underneath a soda straw-type stalactite that had a consistent drip-collecting apparatus placed in Soreq Cave in July 1990. The stalagmite grew underneath a soda straw-type stalactite that had a consistent drip-rate at this site since 2009 (Bursyn, 2013). Sample 5–3b was moved from Soreq in August 2008, and then cut in half along its vertical growth axis. The cross section of 5–3b prepared for analysis measures 5.4 mm along the vertical axis and 10.0 mm across the base. Fig. A.1 in the Supplementary online material shows optical imagery of 5–3b.
before and after being sectioned for casting in a 25 mm-diameter epoxy mount. The 25 mm epoxy round, which includes 3 grains of UWC-3 (calcite standard, $\delta^{18}O$ = 12.49‰, VSMOW; Kozdon et al., 2009), was ground and polished at UW-Madison using diamond paste in a progression from 6 to 1 μm grits. The final polish was attained with a colloidal alumina (0.05 μm) solution. Following the casting and polishing processes, in situ X-ray diffraction analyses with 100 μm spot-size at UW-Madison showed that 5–3b is predominantly calcite; no aragonite was detected (Fig. A.2). Further sample preparation details are described in Orland et al. (2012).

2.1. Confocal laser fluorescent microscopy

Following polishing, 5–3b was imaged by confocal laser fluorescent microscopy (CLFM) at the UW-Madison Keck Bioimaging Lab with a Bio-Rad MRC-1024 scanning confocal microscope. Fluorescence was stimulated with a 40 mW laser (488 nm wavelength), and a filter isolated the emitted wavelengths between 505 and 539 nm — the green portion of the visible spectrum. A series of overlapping images were collected with a 10 × objective lens (100 × total magnification, 959 × 1199 μm field of view) in order to generate a stitched map of the entire sample surface. Because we are unaware of a method to standardize fluorescence intensity during CLFM imaging, the focal-plane depth, laser power, and signal gain were held constant in order to normalize intensity measurements; for this reason, interpretation of fluorescence variability is limited to relative — not absolute — intensity values from the CLFM images. Along with reflected-light images at the same scale, the CLFM map was used to identify growth bands and target spots for $^{83}O$ and trace element analysis by ion microprobe.

2.2. Ion microprobe analysis

Ion microprobe analysis of $^{83}O$ in 5–3b followed the methodology established in earlier work using the CAMECA ims-1280 at the WiscSIMS lab, UW-Madison (Orland et al., 2009, 2012). Before analysis, the epoxy mount was cleaned in two successive 30 s sonication sessions, the first in ethanol and the second in distilled water. Next, a gold coat of ~60 nm thickness was applied to the polished surface to alleviate sample charging during ion microprobe analysis. A focused primary beam of $^{133}Cs^+$ ions was used to sputter each analysis pit in situ. The $^{83}O$ analysis pits were 1 μm deep and oval-shaped, with the long dimension varying for different sessions from 10 to 15 μm depending on both primary-beam tuning and Cs-source conditions. Both $^{81}O^-$ and $^{18}O^-$ ions were collected by Faraday cups in multicollection mode. Using the standard-sample-standard bracketing technique outlined by prior workers (Kita et al., 2009; Kozdon et al., 2009; Valley and Kita, 2009), the average spot-to-spot reproducibility for the $^{83}O$ analyses of UWC-3 calcite standard run with sample 5–3b was ± 0.30‰ (2 standard deviations, s.d.; see the Supplementary material, Table A.1).

This study is the first to report a suite of trace element concentrations in a carbonate from the WiscSIMS ion microprobe. We report elemental concentrations of Mg, Si, P, Mn, Fe, Zn, Sr, Y, and Ba (Table A.2). A 1.0 nA primary beam of $^{16}O^-$ ions with an accelerating voltage of 13 kV was generated from a duoplasmatron source and focused onto the sample surface. Sputtered secondary ions with a positive charge were accelerated across a 10 kV potential into the double-focusing mass spectrometer. Each spot analysis took 10 min and included: pre-sputtering to ablate the Au coating and stabilize secondary-ion emission (60 s), automatic beam centering (60 s), and five 90 s cycles of the magnet scanning through a range of masses (6.5–138). Non-peak positions were included in each analytical cycle at masses 6.5, 6.8, and 131 in order to allow the magnet to stabilize either after completing a cycle (6.5, 6.8) or when jumping a large gap in mass (131). Each analyzed mass was measured for either 2.0 s (Si) or 2.96 s (all others) followed by a 2.0 s waiting period. Ions of each mass were counted by the center-axis electron multiplier except for $^{40}Ca^+$ ions, which were directed into a center-axis Faraday cup to avoid saturating the electron multiplier. The count rate measured for each element during a cycle was normalized to the count rate of $^{40}Ca$ measured in that same cycle. For $^{24}Mg$, $^{31}P$, $^{55}Mn$, $^{57}Fe$, $^{88}Sr$, $^{89}Y$, and $^{133}Ba$, the average of $^{40}Ca$-normalized count rates from all five cycles was used to calculate elemental concentrations. Decreasing count rates observed in the first 3 cycles of $^{26}Si$ and $^{60}Zn$ measurements, perhaps due to surface contamination, are removed from consideration so that for these elements only the final two cycles with stable count rates were averaged for the concentration calculations.

To avoid variability in matrix effects that would stem from using NIST glass as a standard, we used the UWC-3 calcite standard to calibrate ion microprobe measurements of trace elements in 5–3b.

We established the trace element profile of UWC-3 from a combination of electron microprobe analyses at UW-Madison (see Supplemental material in Kozdon et al., 2009), LA-ICP-MS analyses at The Hebrew University-Jerusalem (HUJI; Table A.3), and ICP-atomic emission spectrometry (AES; Perkin-Elmer Optima 5300) and ICP-MS (Perkin-Elmer ELA 6100) analyses by a commercial laboratory (Table A.3). The commercial laboratory analyzed UWC-3 and four working standards (SV-4, Natural Resources Canada; NBS-888, US National Bureau of Standards; BCS-368 and GXR-3, Flanagan (1986)) by sodium peroxide fusion and cites an analytical precision of 10%. We report concentrations of nine elements in UWC-3: three from electron microprobe (Mg, Mn, Fe) and six from LA-ICP-MS (Si, P, Zn, Sr, Y, Ba). Results from the commercial laboratory show excellent agreement with these concentrations (Table A.4). Relative sensitivity factors (RSF) for WiscSIMS analyses of these nine trace elements in calcite were calculated using the average of five UWC-3 analyses at the beginning of the ion microprobe session (Table A.4). The RSF values are used to convert elemental count rates measured in 5–3b into elemental concentrations.

Following the analytical sessions, a thin gold coat was added to the sample surface to allow imaging of the pit-bottoms by scanning electron microscope (SEM). Ion microprobe pits from both trace element and $^{83}O$ analyses were examined at high magnification (> 2500 × ) for irregularities like inclusions, cracks, or epoxy. Each pit-bottom was classified as “regular,” “intermediate,” or “irregular” (Tables A.1 and A.2). For trace element analyses, the irregular pits (n = 3) do not appear to have systematic offsets for any element. Therefore, all trace element analyses are included in our interpretations.

Oxygen isotope analyses are screened in two different ways. First, analyses with anomalous secondary-ion yields are excluded (Kozdon et al., 2009, 2011; Orland et al., 2012). As with the determination of spot-to-spot reproducibility, a standard-sample-standard technique is used to calculate the “percent yield” of each $^{83}O$ analysis relative to the bracketing standard measurements. Analyses with anomalous ion yield values relative to other pits measured in the same session are assumed to reflect a sample abnormality (e.g. a non-carbonate inclusion). Because the range of “percent yield” values for each analysis session is dependent on the primary beam condition, an accepted range of percent yield is defined for each session using the Tukey outlier definition (Tukey, 1977). The accepted range of percent yield in the first analytical session was 94.3–99.3% and 91.1–101.5% in the second analytical session. A total of 12 $^{83}O$ analyses fall outside of the secondary yield cutoff values (Table A.1). These analyses, which represent less than 6% of the 231 sample analyses, are excluded from further consideration.

Oxygen isotope analyses are then screened by SEM imaging. A total of 14 $^{83}O$ pits classified as irregular (Table A.1) are removed from the plots and discussion presented below.

2.3. LA-ICP-MS analysis

LA-ICP-MS analysis of 5–3b was completed at HUJI using a New Wave Research UP-193FX laser ablation system with a custom-built, He-purged “super-cell” (Divir and Kessel, unpublished results). The HUJI laser-ablation set-up is coupled with an Agilent Technologies 7500 series ICP quadrupole mass spectrometer for high-precision measurement.
measurements of elemental concentration. The analytical traverse is comprised of two co-axial laser passes, the first with a low-power (50% laser power, 30 Hz, 10 μm/s stage speed) 100-μm-diameter beam that removed ~5 μm of sample to “pre-clean” the surface. The second pass, for trace element analysis, used a higher power (65% power, 10 Hz, 4 μm/s stage speed) 55-μm-diameter beam that removed a further 10 μm of sample (Fig. A.3). Elemental counts in the ICP-MS were totaled 3 times so that each data point represents an integration of elemental counts from the 55 μm laser spot as it advanced ~1.3 μm. Imaging of an abruptly-terminated laser traverse by reflected-light profilometer (Fig. A.4) indicates that laser ablation rate is relatively consistent across the target such that material at the trailing edge of the laser is ablated at approximately the same rate as the leading edge. Thus, the resulting traverses can be likened to a 55 μm moving-average calculated at 1.3 μm intervals.

Concentrations of 11 elements (23Na, 24Mg, 28Si, 31P, 55Mn, 63Cu, 66Zn, 88Sr, 138Ba, 208Pb, and 238U) were calculated in sample 5–3b using five bracketing analyses – two before and three after the traverse – of a chip of NIST-610 glass standard mounted adjacent to the sample in the super-cell. Each NIST-610 analysis is comprised of a ~30 s (120 μm-long) traverse using the same analytical settings (no pre-clean) as described above. Data reduction was completed with the Iolite software package using 28Ca as an internal standard (Hellstrom et al., 2008; Paton et al., 2011). Since a characterized carbonate standard did not exist, no matrix correction was made to the concentration values measured by LA-ICP-MS (Fig. A.5). Table A.3 suggests that no matrix correction need, because glass-standardized LA-ICP-MS measurements of UWC-3 show excellent agreement with the ICP-AES/MS results.

Since the environmental interpretations presented in this study are primarily based on patterns of trace element variability, it is important to show that relative concentration values measured by LA-ICP-MS across sample 5–3b are reproducible. We approach this issue in two ways. First, we demonstrate analytical reproducibility by calculating the 2 s.d. of elemental concentrations measured in the five bracketing NIST-610 analyses. Using 29Si as an internal standard, the 2 s.d. of Na, Mg, P, Mn, Cu, Zn, Sr, Ba, Pb, and U concentration variabilities measured during the five NIST-610 analyses are <1% of the respective average concentrations (Table A.5). Second, we analyzed another LA-ICP-MS traverse parallel to the analytical traverse, offset laterally by 125 μm and without a pre-cleaning trench. The parallel traverse verifies both the elemental concentration and variability measured in the analytical traverse (Fig. A.6).

### 2.4. Principal component analysis

The trace element results from the continuous laser traverse are well-suited for principal component analysis (PCA; Mix et al., 1986; von Storch and Zwiers, 1999). PCA evaluates the time-series components of empirical orthogonal functions, which reveal common variability in multiple, concurrent datasets. PCA was performed with MATLAB (2010) software using the built-in princomp function. We applied the PCA technique to z-score normalizations (data normalized to have mean = 0, s.d. = 1) of trace element transects and a profile of fluorescence intensity that was measured parallel to the laser trench.

The fluorescence intensity data used in the PCA was obtained from a group of stitched CLFM images that follow the LA-ICP-MS traverse. The fluorescence profile was generated in ImageJ, an open-source image processing software (Rasband, 2012), by averaging the pixel brightness intensity (on a grayscale from 0–255) of a 5-pixel-wide traverse that is offset 75 μm from the laser trench. We used the AnalySeries software (Paillard et al., 1996) to re-sample the fluorescence profile by linear interpolation in order to match the sampling resolution of the LA-ICP-MS data, a requirement for PCA.

### 3. Results

#### 3.1. Fluorescent banding

Fig. 3 shows CLFM imaging of the polished analytical surface – a vertical section – of stalagmite 5–3b. Concentric fluorescent banding is evident along the entire analytical traverse (Fig. 4), which is aligned parallel to the vertical growth axis. There is a striking transition from distinct, bright fluorescent banding in the older (“inner”) portion of the sample to subtle fluorescent banding in the more porous younger (“outer”) portion. The crystal fabric of the stalagmite also changes between the inner and outer portions. Calcite crystals in the inner portion are blocky with low porosity, while the outer portion is highly porous with a dendritic calcite fabric that radiates away from the growth surface.

A discontinuity between the inner and outer portions is highlighted by a dashed line in Fig. 3 and appears to be a healed fracture. Non-fluorescent calcite crystals that are interspersed with voids (~100 μm in diameter) act as infill above the dashed fracture. Laterally continuous growth bands cap the infill and mark the transition to subtle fluorescent banding in the outer portion. The white bracket in Fig. 3 shows an area of “indistinct” banding where it is difficult to trace continuous fluorescent bands. The indistinct banding occurs immediately before the abrupt transition to the subtle fluorescent banding regime.

#### 3.2. δ18O analyses

Values of δ18O measured by ion microprobe span a range of 5.1‰, from 22.5 to 27.6‰ (VSMOW). The average spot-to-spot precision measured for UWC-3 bracketing standards is ±0.30‰ (2 s.d.). Fig. 4A illustrates the location of 213 δ18O analyses made on sample 5–3b (white circles). Fig. 4B shows the δ13C values of each measurement plotted against distance (projected on the LA-ICP-MS traverse). Tabulated results of each ion microprobe δ13C analysis are reported in the data repository, Table A.1. Fig. A.7 is a high-resolution map of the analytical traverse with each ion microprobe analysis labeled.

Seven analysis pits were targeted along a single fluorescent band (Fig. 3) in order to confirm the consistency of δ13C values measured along a single band. As in Orland et al. (2012), along-band analyses are used to assess the influence of variable porosity on the reproducibility of δ13C measurements. We chose to target a distinctive fluorescent band (Fig. 3) in the outer portion of the sample. The seven δ13C measurements along the targeted band in 5–3b have a 2 s.d. of 0.31‰, nearly identical to the 2 s.d. of the bracketing UWC-3 standard (0.30‰). Furthermore, their secondary ion yields vary from 96–98% of the bracketing standards, well within the yield-cutoff values of 94.3 and 99.3% for that analytical session. These results demonstrate the along-band continuity of δ13C values in sample 5–3b.

#### 3.3. Trace element analyses

Trace element concentrations from both ion microprobe and LA-ICP-MS analyses are shown in Fig. 4 for four elements: Mg, Sr, Si, and P. Fig. A.5 includes trace element profiles for all 11 elements measured by LA-ICP-MS. Concentration data from the 42 ion microprobe analyses of trace elements are reported in Table A.2.

We note that the mass used to measure Mn concentration by LA-ICP-MS (54Mn) has a common isobaric interference in ICP mass spectrometers (54Ar31P+) is formed in the Ar plasma; Jochum et al., 2012). LA-ICP-MS measurements of Mn in sample 5–3b are validated by comparable ion microprobe results. Since the calibration of ion microprobe measurements of Mn in 5–3b is based on electron microprobe analysis of Mn in UWC-3 (Kozdon et al., 2009), the ion microprobe measurements of Mn in sample 5–3b provide independent confirmation of the LA-ICP-MS transect.
highlights a growth discontinuity, which appears to be a healed fracture (see text, Section 3.1). The rectangle outlines the area of sample 5 displayed in Fig. 4. Yellow hashes mark the annual growth increments as interpreted in Fig. 4. The white bracket shows the location of the indistinct banding that occurs immediately band (Section 3.2). Note that the growth rate of 5-3b is not a linear function of annual band widths along the analytical traverse because the analytical surface bisects a half-spherical sample.

3.4. Principal component analysis and correlation results

The PCA results illustrated in Fig. 5 and discussed below incorporate 12 variables. They are: (1) a record of fluorescence intensity across 5–3b as measured by CLFM and (2) 11 LA-ICP-MS elemental transects with signals discernable above background noise (Na, Mg, Si, P, Mn, Cu, Zn, Sr, Ba, Pb, U).

PCA results are displayed in Fig. 5 in three manners: the “percentage of total variance” explained by each principal component, the sample “scores”, and the variable “loadings.” We briefly outline their meaning here. PCA redistributes the variance observed in our 12 variables into 12 orthogonal axes called “principal components” (PC) that are oriented according to patterns of co-variation in the original data (Davis, 2002). In concept, the principal components reflect underlying forcing mechanisms. Each of the PCs describes a “percentage of the total variance” observed in the original data, and they are ranked according to this value (i.e. the first principal component, PC1, explains the highest percentage of total variance). Each PC is described by a time-series of normalized “scores” that defines a pattern of variability that is unique to that principal component. For every PC, all 12 of the input variables are assigned a “loading” on a scale from −1 to 1 that indicates how well each variable correlates to the PC. A loading value of one denotes perfect positive correlation to a PC, while a negative loading value signifies an inverse correlation. In Fig. 5D–F, the loadings are plotted for each element relative to the first three PCs. Different classes of elements with shared geochemical properties are outlined in color to ease visual comparison. Alkaline earth metals are highlighted in blue (Mg) and red (Sr, Ba), while colloid- and silicate-borne elements are highlighted in green (P, Cu) and yellow (Si, Mn), respectively.

The percentage of total variance explained by the first three PCs are 26%, 24%, and 14%. Together, they yield a cumulative percentage of 64% of the total variance explained by all of the PCs. For each of the nine remaining PCs, the percentage of total variance is <8% and as such, these PCs are not discussed below as significant components of the LA-ICP-MS dataset.

Two tests verify the patterns of variability illustrated in Fig. 5 (Fig. A.8). First, we note that the large peaks in the scores of PCs 1–3 (Fig. 5A–C) correspond to high trace element concentrations near the upper edge of the indistinct layer (e.g. Sr; Fig. 4B) as well as at the very top of the sample. PCA of the LA-ICP-MS results minus the portions with exceptional trace element concentrations gives similar PC scores and elemental loadings to those illustrated in Fig. 5. The second test aims to assess the impact of the relatively wide band at the base of the sample on the PC patterns by removing the first 1000 μm of the analytical traverse from the PCA. Although this represents a 20% decrease in the analytical traverse, because there is little elemental variability in this section (see Fig. 4) the resulting PC scores and elemental loadings are again similar to those in Fig. 5.

In addition to PCA, a combination of three statistical conditions is used to identify significant correlations among geochemical variables in sample 5–3b; (1) Pearson r-value > 0.5, (2) respective p-value < 0.02, and (3) Spearman rank correlation r-value > 0.5. Both Pearson r-values and Spearman p-values are measured on a scale from 0 to 1 with higher values indicating increased correlation. The p-value indicates the probability on a scale from 0 to 1 that a higher value of Pearson’s r-value could be obtained from a non-correlating dataset. Thus, low p-values indicate more significant r-values.

Pearson correlation coefficients (r-values; Johnson and Bhattacharyya, 2009) for the inner and outer 22 trace element pits are tabulated in Table 1; correlations that meet all three conditions from above are given in bold. The dataset was subdivided into three portions (inner, indistinct, outer) based on the fluorescent banding in Fig. 3. The four pits analyzed within the indistinct banding layer are excluded from the correlations in Table 1; these four pits only served to mask correlation coefficients of the inner and outer datasets. In order to assess the correlation of each trace element to δ18O, we identified the δ18O pit closest to the location of each trace element analysis. Although Fe and Zn concentrations were measured by ion microprobe, their respective correlation values are not included in Table 1 because they never meet the three significance conditions. All three statistical results for both ion microprobe and LA-ICP-MS analyses of sample 5–3b are available in Table A.6.
4. Discussion

Earlier work compared δ¹⁸O values measured by ion microprobe to fluorescence imaging by CLFM in two Soreq speleothems (34–4 ka and 2.2–0.9 ka) and proposed that: (1) fluorescent bands in the speleothems represent annual growth bands, and (2) fluorescence and δ¹⁸O values observed within single annual growth bands record seasonal rainfall patterns like the distinct wet and dry seasons of the modern climate regime (Orland et al., 2009, 2012). The suite of fluorescence, δ¹⁸O, and trace element results from modern sample 5–3b test these hypotheses and add to our understanding of micro-scale trace element variability within Soreq speleothems. This information will help to calibrate paired micro-analyses of δ¹⁸O and trace elements in future investigations of speleothems from semi-arid environments.

4.1. Chronology of sample 5–3b

4.1.1. Fluorescent banding

The first step towards determining if fluorescent bands are annual is to count the number of fluorescent bands. Since the sample substrate was placed in the cave in July 1990 and sample 5–3b was collected in August 2008, we expect to see 18 annual bands assuming that all years are recorded.

In Fig. 3, we find 12 bands (marked by the longer yellow hashes) that are distinguished by laterally-continuous fluorescent boundaries. Identifying additional definitive bands from just the CLFM imaging is challenging; the six bands marked with shorter hashes in Fig. 3 are interpreted as annual based on further insight gained from the geochemical analyses discussed below. Therefore, although fluorescent imaging identifies a
majority of the annual growth boundaries in 5–3b, geochemical analyses are needed to construct an 18-year age model. Notably, the composite fluorescent image in Fig. 3 also shows that fluorescent bands in the inner portion of sample 5–3b are characteristically more intense than bands in the outer portion. Two interpretations are possible for the change in fluorescence. First, it is conceivable that whatever disturbance caused the discontinuity (dashed line in Fig. 3) also repositioned the man-made substrate so that sample 5–3b was directly beneath a new drip-point that inhibited fluorescence. Such changes are unlikely, especially since the vertical growth axis appears unchanged. Instead, we prefer a second hypothesis that the intense fluorescent bands indicate a wetter period when dripwaters above 5–3b carried an increased load of fluorescent material into the cave from the overlying soil. Correspondingly, the subtle banding in the outer portion indicates a drier period when either the supply or delivery of fluorescent material into the cave was suppressed. Geochemical analyses of both sample 5–3b and rainwater collected above the cave allow us to scrutinize the suggested annual bands and test our explanation for the reduction in fluorescence intensity.

4.1.2. Anchor points from the $\delta^{18}O$ record

At Soreq, Ayalon et al. (1998) show that the seasonal $\delta^{18}O$ signal in rainfall above the cave is imparted to dripwaters. This section further describes how the local rainfall $\delta^{18}O$ signal between 1990 and 2008 (Table A.7; Ayalon et al., 1998; Burg, 1998; Ayalon et al., 2004; Burstyn, 2013) is conveyed to Soreq speleothems and identifies prominent features in the rainfall record that can be used to anchor the chronology of 5–3b. Fig. 2 illustrates the strong seasonal gradient of both rainfall amount and $\delta^{18}O$ values measured above Soreq Cave from 1995–2008; $\delta^{18}O$ values are lowest in the winter wet-season and highest in the fall and late spring. Since >95% of rainfall occurs between November and April,

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Fig. 5. PCA results for 12 datasets that traverse sample 5–3b, including 11 records of trace element concentration from LA-ICP-MS analysis (Na, Mg, Si, P, Mn, Cu, Zn, Sr, Ba, Pb, U) and one record of fluorescence intensity acquired by CLFM. Section 3.4 explains the PCA terminology. Panels A–C show the scores of PCs 1–3, respectively, plotted versus distance from the base of 5–3b. The percentage of total variance (%) represented by each PC is indicated in the upper-left corner of panels A–C. Panels D–F show the loadings of the 12 observed datasets for PC2 vs. PC1, PC3 vs. PC1, and PC3 vs. PC2, respectively. Since the sum of the squared loading values of a trace element on all of the principal components equals 1, the unit circles in panels D–F outline the maximum range of possible loadings. For example, if one trace element has a loading of 1 on any principal component, it must have a loading of zero on all others. Notable classes of elements are highlighted in color to facilitate comparison of their relative loadings; divalent cations are colored blue (Mg) and red (Sr, Ba), while elements transported by organic colloids or silicates are green (P, Cu) and yellow (Si, Mn). The similar variability of fluorescence intensity (CLFM) to that of P and Cu likely indicates that fluorescence in sample 5–3b is caused by P- and Cu-laden organic colloids. (For interpretation of the references to color in this figure legend, the reader is referred to the web version of this article.)
we use the notation “water year” to delineate wet seasons; a water year extends from October of one year through September of the next and is designated by the calendar year in which it ends. Fig. A.9 shows how the cumulative \( \delta^{18}O \) of rain from each water year tracks the \( \delta^{18}O \) of rainfall from that wet season. Thus, to simplify the presentation of rainfall data in both the tables and figures of this discussion, we reference values of rainfall amount and \( \delta^{18}O \) integrated over entire water years.

Fig. A.9 also includes \( \delta^{18}O \) values measured in dripswaters collected near sample 5–3b between 1991 and 1997 and starting again in 2001. These data (tabulated in Table A.8) have two important features that demonstrate how the rainfall \( \delta^{18}O \) signal is transmitted into the cave. First, driprwater \( \delta^{18}O \) is lowest during the wet season, as also observed in rainwater above the cave. Second, the minimum driprwater \( \delta^{18}O \) values from each year track with annual rainwater \( \delta^{18}O \). In a cave-wide driprwater survey, Ayalon et al. (1998) found that minimum driprwater \( \delta^{18}O \) values each year were consistently >1‰ higher than the annual rainfall \( \delta^{18}O \) value. Fig. A.9 indicates a similar offset for driprwaters collected after 1998.

Critical for this study is the assumption that \( \delta^{18}O \) values measured in sample 5–3b reflect near-equilibrium of \( \delta^{18}O \) between calcite and driprwater. Recent modeling (Deininger et al., 2012) and experimental (Gabitov et al., 2012; Riechelmann et al., 2013) studies of isotope disequilibrium in speleothums suggest that seasonal variability of drip- and growth-rates could act to amplify the seasonal gradient of speleothum \( \delta^{18}O \) relative to Soreq driprwater. Given the relatively low drip-rates in Soreq, however, these effects would impart <1‰ to the seasonal gradient — not enough to explain the observed variability. Furthermore, measurements of mass–47 CO\(_2\) isotope analogue anomalies (\( \Delta_{47} \)) in modern Soreq speleothums indicate that modern calcite precipitated in near-equilibrium conditions for \( \delta^{18}O \) although \( \Delta_{47} \) values are not fully equilibrated (Affek et al., 2008). This accords with an earlier study of calcite precipitation in modern cave pools (Bar–Matthews et al., 1996).

With the knowledge that rainwater \( \delta^{18}O \) variability is reliably transmitted to Soreq speleothums, two water years in the rainfall record stand out as being useful for correlating with the 5–3b \( \delta^{18}O \) record (Figs. 6 and A.9). The first is 1992, which had the highest annual rainfall amount over the period of interest (1020 mm) and the lowest \( \delta^{18}O \) value of cumulative annual rainfall (–8.0‰, VSMOW). The second notable year is 1999, which had the lowest rainfall total of 215 mm and the highest cumulative \( \delta^{18}O \) value, –4.5‰. Fig. 6 plots the rainfall totals and cumulative rainwater \( \delta^{18}O \) values from 1991–2008 as well as all \( \delta^{18}O \) values measured in 5–3b by microprobe. As in Fig. 4, the horizontal axis of Fig. 6 reflects the linear distance along the LA-ICP-MS traverse. We position the 1992 and 1999 water years (indicated with thick vertical dashed lines in Figs. 4 and 6) to match the portions of 5–3b with extreme \( \delta^{18}O \) values.

In order to identify other years in the chronology, we look for the pattern of sawtooth \( \delta^{18}O \) variability that earlier studies characterized in annual growth bands from Holocene Soreq speleothums (Orland et al., 2009, 2012). Using this approach, wet-season markers are assigned to the outer three growth bands (2006–08) as well as the four bands (1993–96) that follow the particularly wet 1992 water year (Fig. 4). In the intervening portion of the sample, which represents growth between 1997 and 2005, sawtooth patterns of \( \delta^{18}O \) are apparent but fluorescent bands are less definitive. Therefore, trace element results are examined for complementary evidence of annual growth bands.

### Table 1

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<th>( \delta^{18}O )</th>
<th>Mg</th>
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**Table A.6**

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4.1.3. Trace element variability and chronology

Many studies have identified records of annual trace element variability in speleothems (Fairchild et al., 2001; Treble et al., 2003, 2005; Johnson et al., 2006; Borsato et al., 2007; Fairchild and Treble, 2009). Together they provide valuable insight on the source and pathways of trace elements analyzed in sample 5–3b. For example, recent work (Borsato et al., 2007; Hartland et al., 2012) shows that P, Cu, Zn, Y, Pb, and other trace metals are commonly carried into the cave as adsorbed species on colloids, which likely includes organic particles from the overlying soil. The trace elements most commonly measured at subannual resolution in speleothem studies are the alkaline earth metals, Mg and Sr. They form divalent cations in solution, substitute for Ca$^{2+}$ in calcite, and are particularly useful geochemical indicators in caves—like Soreq—which formed within dolomite bedrock (Table A.9).

Variability of Mg and Sr in dripwater is often attributed to changes in the amount of either groundwater–bedrock interaction or prior calcite precipitation (Fairchild and Treble, 2009; Schimpf et al., 2011; Wong et al., 2011). In drier conditions, the concentration of Mg and Sr in groundwater can be increased relative to Ca because: 1) longer periods of groundwater–bedrock interaction add higher proportions of Sr and Mg to the groundwater, or 2) calcite precipitation in rocks above the cave preferentially removes Ca from groundwaters. However, rather than attributing Mg and Sr variability in sample 5–3b to the seasonal influence of either or both of these mechanisms, we adopt the hydrological routing model of Fairchild et al. (2006) to explain trace element variability. They suggest that Mg and Sr concentrations in dripwater represent a mixture of groundwaters that have experienced different residence times or flow pathways within the vadose zone. This is the same model used to explain $\delta^{18}$O variability in Soreq Cave dripwaters and speleothems (Ayalon et al., 1998; Kaufman et al., 2003; Orland et al., 2009).

Fig. A.9 shows that Mg concentrations in Soreq dripwaters (tabulated in Table A.8) adhere to the suggested model, with generally increasing concentrations following each wet season. Before using both Mg and Sr concentrations to identify annual bands in 5–3b, however, we take into account that: (1) in Soreq Cave speleothems, wind-blown dust is known to be a source of Sr (Ayalon et al., 1999; Bar-Matthews et al., 1999; Frumkin and Stein, 2004), and (2) Sr and Mg variability do not match in the inner portion of 5–3b (Fig. 4). Therefore, we focus on Mg variability to guide identification of the remaining annual bands in 5–3b.

Much like with $\delta^{18}$O, we interpret a sharp decrease in Mg concentration to indicate the onset of annual wet seasons; following the sharp decrease, a gradual increase in Mg concentration is expected as the relative contribution of longer-residence time groundwater grows. This interpretation corroborates the delineation of the 2006–08 growth bands identified by $\delta^{18}$O variability. Furthermore, sawtooth variability of Mg is evident (although smoothed by the continuous laser traverse) in growth bands from 2000–05 where fluorescent bands were less conclusive. The remaining annual bands, 1997–99, are assigned by sawtooth $\delta^{18}$O patterns that coincide with subtle Mg and fluorescence variability (Fig. 4).

4.2. Comparison of rainfall and speleothem $\delta^{18}$O records

If we assume the age model constructed in the previous section is accurate, we can test the hypothesis of Orland et al. (2009) who suggest that the range of $\delta^{18}$O in each band of a Late Holocene speleothem is a proxy for rainfall amount. They define a variable, $\Delta^{18}$O = $\delta^{18}$O(dark calcite) – $\delta^{18}$O(bright calcite), to describe the $\delta^{18}$O range in individual growth bands. Here, since annual growth bands are not defined solely on the basis of fluorescence, $\Delta^{18}$O values are denoted as $\Delta^{18}$O$_{3b}$ and refer to the range of $\delta^{18}$O measured within the $\delta^{18}$O sawtooth recorded in each growth band. Fig. 6D plots the $\Delta^{18}$O$_{3b}$ value calculated for each band in 5–3b. Fig. 7 illustrates a very good correlation between $\Delta^{18}$O$_{3b}$ values measured in 5–3b and annual rainfall totals from above the cave ($r^2 = 0.75$), and thus supports the hypothesis that $\Delta^{18}$O$_{3b}$ values are a proxy for annual rainfall amount in Soreq speleothems.

The correlation between $\Delta^{18}$O$_{3b}$ and annual rainfall amount provides further insight into the two-reservoir groundwater system outlined above. According to the model of annual $\delta^{18}$O variability described by Orland et al. (2009), the minimum $\delta^{18}$O value of calcite in each annual band should approximately reflect the cumulative $\delta^{18}$O of rainfall from the respective water year. In sample 5–3b, these values correlate with $r^2 = 0.42$. Similarly, the correlation between the minimum calcite $\delta^{18}$O in each band and annual rainfall amount ($r^2 = 0.37$; Fig. A.10) is somewhat weaker than that between $\Delta^{18}$O$_{3b}$ and annual rainfall amount ($r^2 = 0.75$, Fig. 7). If our age model is correct, we suggest that the stronger correlation of $\Delta^{18}$O$_{3b}$ and annual rainfall amount is a result of some mixing between the short- and long-residence-time groundwater reservoirs that is weighted by the amount of wet-season rainfall, its $\delta^{18}$O value, and the time since it fell above the cave.

4.3. Climate control of seasonal trace element variability

Building from the chronology of 5–3b proposed above, this study aims to identify signatures of seasonal climate within sub-annual trace element variability. Annual rainfall totals measured between 1991 and 2008 provide a benchmark against which to compare the geochemical data from sample 5–3b. In addition to the extreme wet and dry events used to anchor the chronology, the average amount of rainfall during a water year drops 195 mm from 628 to 433 mm between 1991–98 and 1999–2008. Ideally, the patterns of sub-annual geochemical variability reflect the shift from a wetter to a drier decade. One encouraging observation is the change in the character of fluorescent banding from the younger to older portion of sample 5–3b (Section 3.1); the distinctive fluorescence regimes likely reflect a change in dripwater chemistry that we can quantify with trace element analyses.

There is good agreement between ion microprobe and LA-ICP-MS analyses of Mg, Sr, Si, and P concentrations in sample 5–3b (Fig. 4). Similar plots for Ba, Mn, and Zn (Fig. A.5) also show agreement between concentration values measured by the two methods. Even before statistical analysis, common patterns of variability are clear within different classes of trace elements; one class is comprised of Mg, Sr, and Ba, another includes Si, Fe, and Mn.

4.3.1. Correlation of ion microprobe data

In their study of trace elements in speleothems, Wassenburg et al. (2012) emphasize elemental correlations with $r$-values > 0.5. Our study uses an additional condition to identify the significant elemental correlations that are highlighted in Table 1; the Spearman rank correlation coefficient ($\rho$-value; Johnson and Bhattacharyya, 2009) must also be > 0.5. By including the $\rho$-value requirement, we eliminate correlations...
whose \( r \)-values are skewed by single extreme measurements (e.g., Mn vs. Fe).

The highlighted correlations in Table 1 are markedly different in the inner and outer portions of sample 5–3b. In the inner portion, the noted correlations include elements that are likely carried into the cave during relatively wet periods as either: (1) species adsorbed to organic colloids (P, Y) that may also cause fluorescence, or (2) detrital silicates, possibly including windblown dust (Sr, Mn, Si). In the outer portion, the notable correlations include elements that reflect the residence time of groundwater in the vadose zone before it reaches the cave (Mg, Sr, Ba, \( \delta^{18}O \)). The only correlation common to both inner and outer portions of the sample is between Sr and Ba, a logical result considering the similar ionic radii, valence states and partition coefficients (Day and Henderson, 2013) of these two species.

The switch in significant elemental correlations matches the change in rainfall measured above the cave and inferred from the shift to subtle fluorescent banding. During the relatively wetter conditions of 1991–98, both the distinct fluorescent banding along with P–Y–Sr and Mn–Si correlations indicate that wet-season drippwater included material washed into the cave from the near-surface. In contrast, when the rainfall record shows relatively dry conditions, subtle fluorescent banding and Mg–Sr–Ba–\( \delta^{18}O \) correlations indicate that drippwater had both a longer residence time in the vadose zone as well as a less direct connection with the near-surface. The difference between the geochemical response recorded in the inner and outer portions of 5–3b is likely caused by the amount of suspended load carried in ground-water; trace elements that are preferentially transported by organic acids, colloids, and small particles – including wind-blown dust – are emphasized during periods of increased rainfall.

4.3.2. Principal components of LA-ICP-MS data

The panels on the left of Fig. 5 show the time series scores of the first three PCs and their respective percentages of total variance explained. The high percentage of variance explained by PC1 and PC2 relative to each of the remaining principal components suggests that PC1 and PC2 reflect the dominant modes of trace element variability in sample 5–3b. The panels on the right of Fig. 5 compare the loadings of 11 trace elements on each PC by plotting the loading of each element on PC2 vs. PC1, PC3 vs. PC1, and PC3 vs. PC2. These figures illustrate how different elemental concentrations display common patterns of variability; elements that plot near one another likely share common forcing mechanisms.

Each of the first three PCs is characterized by high loadings of different groups of elements. PC1 has high loadings of Mg, Sr, and Ba that likely reflect the effect of groundwater–bedrock interaction in the vadose zone. PC2 has high loadings of P, Cu, Sr, and U. These elements readily adsorb to colloids and organic acids (P, Cu) as well as wind-blown dust (Sr, U; Frumkin and Stein, 2004), suggesting that PC2 indicates the amount of particulate or colloidal components that are washed into the cave from the overlying soil. PC3 is the only remaining principal component that explains >8% of the total variance and has a pattern consistent with the inclusion of silicate particles in the indistinct banding layer; the highest positive elemental loadings in PC3 are Zn, Pb, Mn, and Si.

4.3.3. Characterizing groundwater reservoir compositions

Another notable feature of the PCA results is the relation of Mg to Sr and Ba in PCs 1 and 2. Fig. 5D illustrates how Mg, Sr, and Ba have similar positive loadings on PC1, but Mg has a negative loading on PC2 while Sr and Ba have positive loadings. This conflicting pattern is comparable to the correlations seen in the ion microprobe data; strong positive correlations are observed between Mg–Sr and Mg–Ba in the outer portion of 5–3b, but these same correlations are weak in the inner portion (Table 1). We hypothesize that the first two principal components represent the same forcing mechanisms ascribed to the correlation data. As such, PC1 would illustrate the influence of groundwater residence time on Mg, Sr, and Ba concentration, while PC2 reflects the amount of suspended material washed into the cave.

In order to test the similarity of the PCA and correlations results, we conducted PCA on the same inner and outer portions of 5–3b as used in the correlation treatment. The results (Fig. A.11) show that the first PC of the inner portion matches the variability of PC2 from the entire sample (Fig. SB); this pattern is dominated by similar variability in Sr, Ba, Cu, P, and U. The first PC of the outer portion, however, is characterized by similar loadings of Mg, Sr, and Ba and has a comparable pattern of variability to PC1 of the entire sample (Fig. 5A). Thus, we are confident that PCs 1 and 2 of the entire sample represent the same forcing mechanisms used to explain correlations in the outer and inner portions, respectively (Table 1).

Given the proposed forcing factors for PCs 1 and 2, Fig. 5D is particularly interesting because it shows the loading of different elements with respect to both components. One feature that stands out is the strong positive loading of both Sr and Ba (outlined in red) on PCs 1 and 2. In this case, PCA illustrates how the two forcing factors overlap one another in the composite trace element record. On one hand, since Sr and Ba concentrate in groundwater that has decadal residence times in the dolomitic vadose zone, they reflect the proportion of vadose zone groundwater in drippwater (PC1). On the other hand, these elements are also delivered to the sample via small particles (likely wind-blown dust) that are flushed into the cave by intense wet season rains (PC2).

Therefore, trace element data reinforce the two-component hydrological routing model previously used to describe \( \delta^{18}O \) and fluorescence variability in Soreq speleothems (Orland et al., 2009). While mixing of groundwater reservoirs is sufficient to explain the \( \delta^{18}O \) variability of 5–3b, the seasonal delivery of particulate matter is critical to understanding the variability of trace elements like Sr, Ba, U, P, and Cu.

With regard to the influence of organic colloidal particles, we emphasize one final observation from the PCA results. Panels D–F of Fig. 5 illustrate the remarkably similar variability of fluorescence intensity and P and Cu concentrations across sample 5–3b; the loadings of P, Cu, and CLF are grouped together (green outline) in each of the first three principal components. The similar patterns of variability support the hypothesis that organic molecules, which are vehicles for P, Cu, and other elements (Borsato et al., 2007; Fairchild and Treble, 2009; Hartland et al., 2012), cause the fluorescent banding in sample 5–3b.

5. Conclusions

We apply high-spatial-resolution analysis and imaging to a modern Soreq Cave stalagmite in order to evaluate the co-variation of \( \delta^{18}O \) and fluorescence variability in Soreq speleothems (Orland et al., 2009). Given the proposed forcing factors for PCs 1 and 2, Fig. 5D is particularly interesting because it shows the loading of different elements with respect to both components. One feature that stands out is the strong positive loading of both Sr and Ba (outlined in red) on PCs 1 and 2. In this case, PCA illustrates how the two forcing factors overlap one another in the composite trace element record. On one hand, since Sr and Ba concentrate in groundwater that has decadal residence times in the dolomitic vadose zone, they reflect the proportion of vadose zone groundwater in drippwater (PC1). On the other hand, these elements are also delivered to the sample via small particles (likely wind-blown dust) that are flushed into the cave by intense wet season rains (PC2).

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period from 1991–1998, fluorescent bands are distinct and seasonal infiltration of organic colloids and small particles of wind-blow dust appears to accentuate the variability of P, Cu, Sr, Na, and U.

Results support the two-reservoir model of Soreq Cave hydrology described by Orland et al. (2009). The first reservoir is stored in pores in the unsaturated zone where it accumulates divalent cations from the dolomite country rock, has a decadal residence time for $\delta^{18}O$ values in water, and supplies well-mixed “baseline” drips to the cave on a year-round basis. The second reservoir delivers wet season rainfall to the cave along more direct pathways and thus has a shorter, sub-anual residence time. PCA results also indicate that the variability of some trace elements (Sr, Ba, U, P, Cu) is overprinted by the delivery of particulate matter during intense wet season rains.

Similar variability of P, Cu, and fluoride intensity in a PCA of sample 5–3b data supports the hypothesis that fluorescent banding is caused by organic colloids washed into the cave from the overlying soil.

Supplementary data to this article can be found online at http://dx.doi.org/10.1016/j.chemgeo.2013.11.011.

Acknowledgments

The authors thank S. Meyers, J. Williams, and A. Carlson for fruitful discussions; N. Kita and T. Ushikubo for support at WiscSIMS; O. Dvir for assistance with LA-ICP-MS analysis; L. Rodenkirch for assistance at the W. M. Keck Laboratory for Biological Imaging at UW-Madison; D. Ortiz and J. Fournelle for guidance on the SEM; B. Hess for sample preparation; J. Kern for proflmeter assistance; H. Konishi for operating the XRD; A. Pollington and E. Syracuse for MATLAB support; the Israel Nature and Parks Authority for access to Soreq Cave. M. Böttcher and two anonymous reviewers provided helpful comments that improved this manuscript. Funding for this project was provided by the NSF (AGS-1003487, EAR-0839058), the Comer Science and Education Foundation, and the United States–Israel Binational Science Foundation (2010316). WiscSIMS is partially supported by NSF–EAR (0319230, 0744079, 1054366).

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