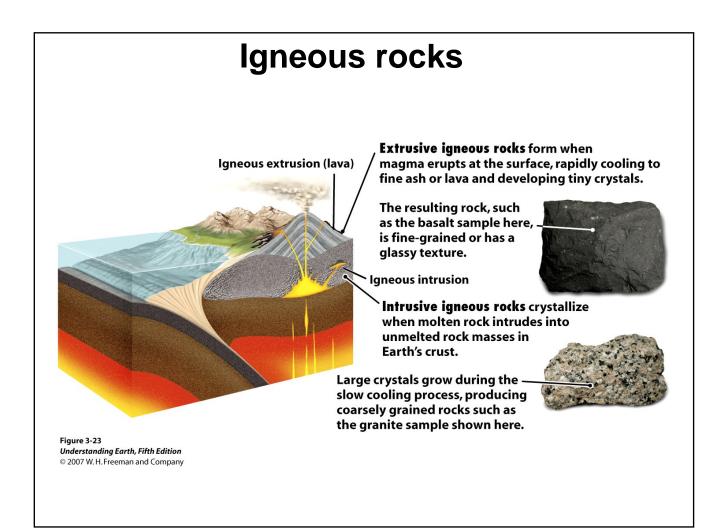
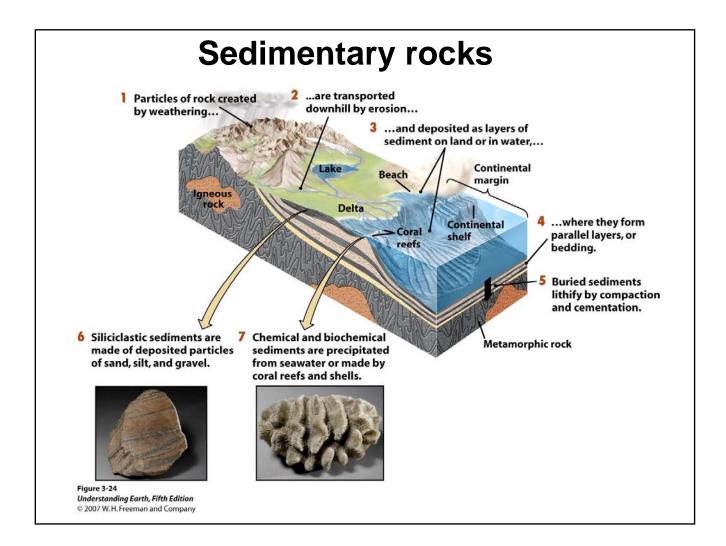
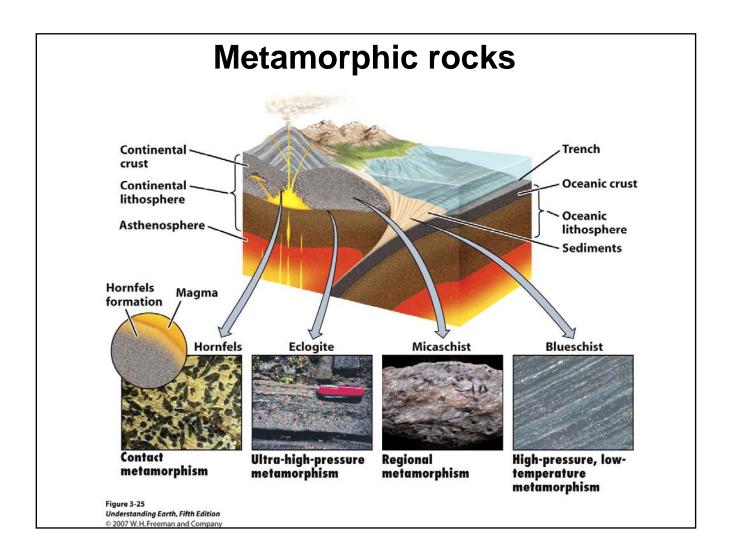


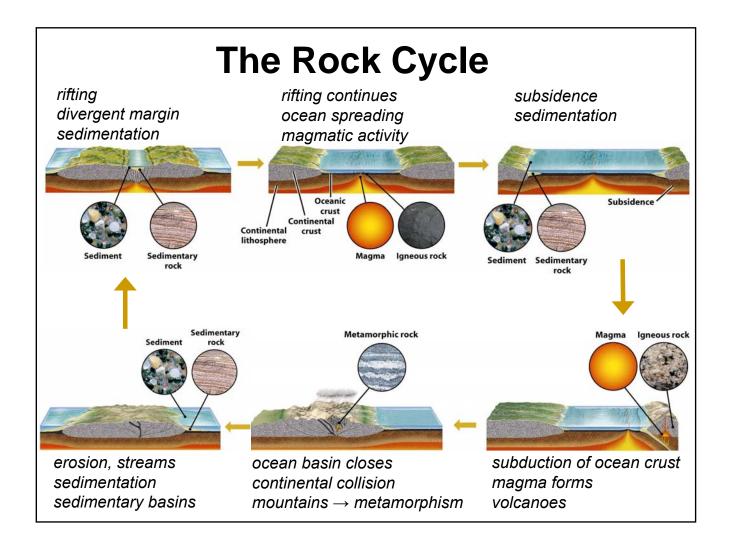
Table 3.5		Some Common Minerals of Igneous, Sedimentary, and Metamorphic Rocks	
lgneous Rocks		dimentary cks	Metamorphic Rocks
*Quartz	*Q	uartz	*Quartz
*Feldspar	*Cl	ay minerals	*Feldspar
*Mica	*Fe	eldspar	*Mica
*Pyroxene	C	alcite	*Garnet
*Amphibole	D	olomite	*Pyroxene
*Olivine	G	ypsum	*Staurolite
	Н	alite	*Kyanite

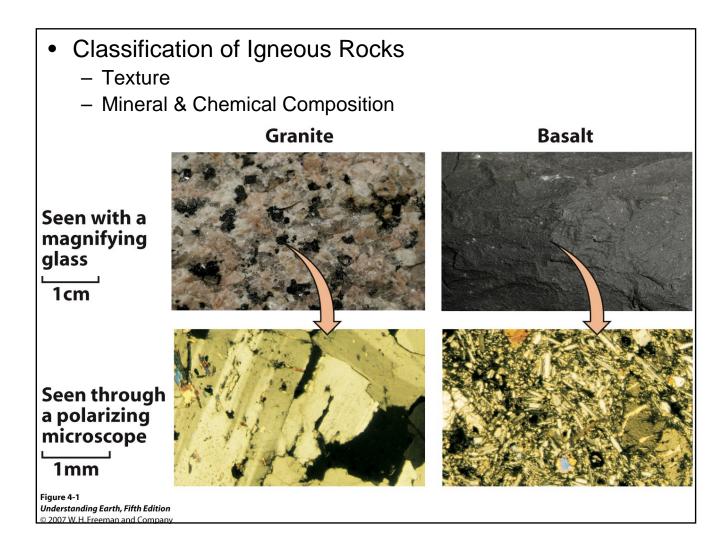
Understanding Earth, Fifth Edition © 2007 W. H. Freeman and Company

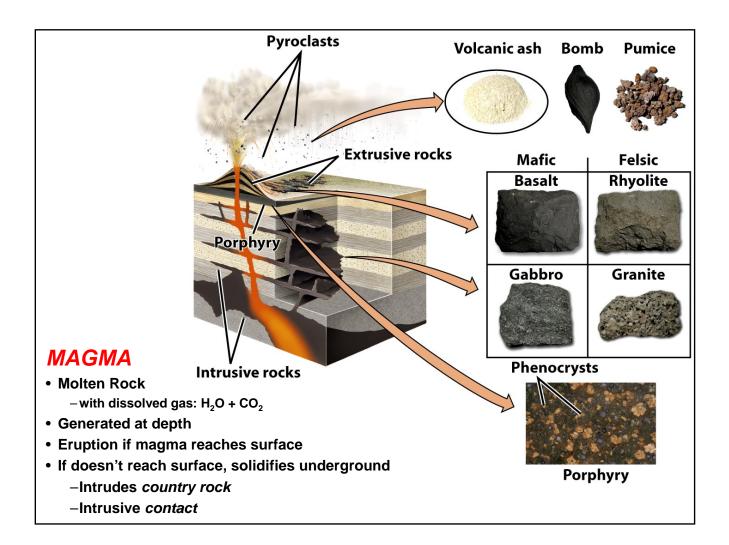




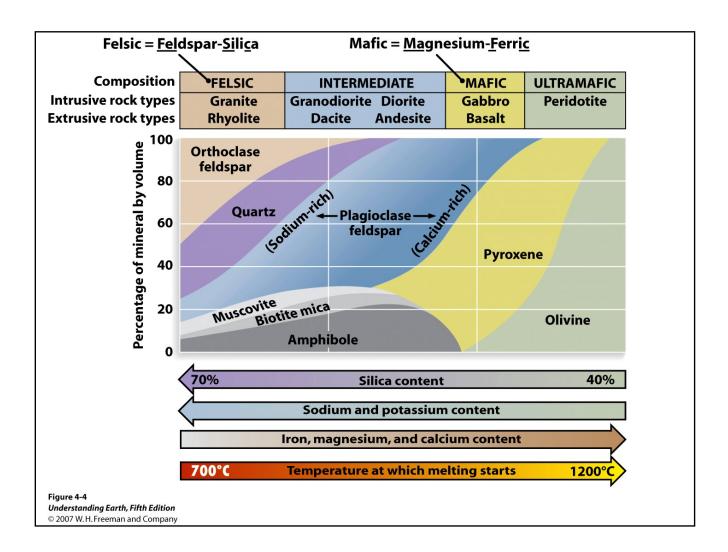


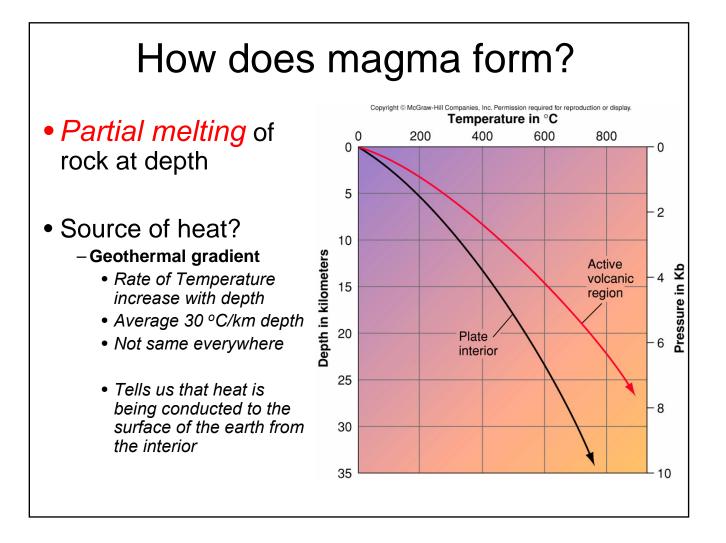






Compositional Group	Mineral	Chemical Composition	Silicate Structure
	Quartz	SiO2	Frameworks
FELSIC	Potassium feldspar	KAISi ₃ O ₈	
	Plagioclase feldspar	NaAlSi ₃ O ₈ ; CaAl ₂ Si ₂ O ₈	
	Muscovite (mica)	KAI ₃ Si ₃ O ₁₀ (OH) ₂	Sheets
MAFIC	Biotite (mica)	K Mg Fe Al	
	Amphibole group	Mg Fe Ca Na	Double chains
	Pyroxene group	Mg Fe Ca Al	Single chains
	Olivine	(Mg,Fe) ₂ SiO ₄	Isolated tetrahedra





How does magma form?

Factors that control melting temperatures

-Pressure

- melting points of minerals increase with pressure
 - This is why increasing temperature along the geotherm alone fails to melt crustal rocks
- reduction in pressure can therefore induce melting

-Water added under pressure

• lowers melting point of minerals



How does magma evolve?

Magmatic differentiation

-Mirror image of partial melting

- Crystal formation over a range in temperature
 - First to melt are also first to crystallize

• Fractional crystallization

 As crystallization proceeds, crystals are separated from the melt, thus the melt evolves away from the composition of the crystals removed

• Bowen's reaction series, a model of crystallization:

